University of Mumbai Examination 2020

Program: Automobile Engineering Curriculum Scheme: Rev 2012 & Rev2016 Examination: Second Year Semester: III Course Code: AEC302 and Course Name: Thermodynamics Time: 1 hour Max. Marks: 50

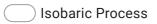
1. In approach, a certain quantity of matter is considered, without a concern on the events occurring at the molecular level.

Mark only one oval.

\bigcirc	macroscopic
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- microscopic
- Menuscropic
- _____ random
- 2. A process in which all states of the system passes through are equilibrium states is called as

Mark only one oval.



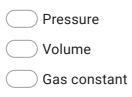


- Polytropic Process
- Adiabatic Process
- 3. Heat transfer is zero in

- Polytropic Process
- Adiabatic Process
- Constant pressure process
- Constant volume process

4. The enthalpy of an ideal gas is independent of its , and depends only on its temperature

Mark only one oval.



- index
- 5. Energy is a of a system.

Mark only one oval.

() State	State
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- Property
- Process
- None of the above
- 6. is a property of a system which determines the degree of hotness or coldness.

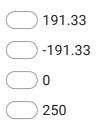
- Pressure
- O Volume
- _____ Temperature
- Viscosity

 In constant volume process the pressure at the initial state was 1 bar and temp was 300 k. At the end state the pressure was rise to 10 bar and temp. was 600K. Find the work done in KJ?

Mark only one oval.

8. Air is compressed poly-tropically till it's temp becomes 500 K. the initial temp of air is 300K. find the work done (KJ/Kg)? index of the process is 1.3

Mark only one oval.



9. Untitled Question

Mark only one oval.

Option 1

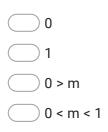
10. What is the relation between efficiencies of Rankine cycle and Carnot cycle for the same pressure ratio?



- (ηRankine) > (ηCarnot)
- (ηRankine) < (ηCarnot)
- ____ (ηRankine) = (ηCarnot)=0

11. For the dry steam, dryness fraction m is =

Mark only one oval.



12. The point of a substance is the temperature and pressure at which the three phases (gas, liquid, and solid) of that substance coexist in thermodynamic equilibrium.

Mark only one oval.

first	
Sec	ond
tripl	е
last	

13. The temperature is the temperature for a corresponding saturation pressure at which a liquid boils into its vapor phase.

- saturation fusion vanishing
- _____ cold

14. The cycle which consists of two reversible isotherms and two reversible adiabatic is called as

Mark only one oval.

Rankine cycle

Carnot cycle

Stirling cycle

Ericsson cycle

15. Find the efficiency of otto cycle having compression ratio 8.

Mark only one oval.

\bigcirc	56.5%
\bigcirc	66.8%
\bigcirc	48%
\bigcirc	25%

16. What is the relation between compression ratio and the efficiency of the Otto cycle?

- efficiency decreases with increase in compression ratio
- efficiency increases with increase in compression ratio
- efficiency does not affected by change in compression ratio
- _____ efficiency may increase or decrease by change in compression ratio

17. How is the heat added in the Dual cycle?

Mark only one oval.

- reversibly partially at constant volume and partially at constant pressure
- irreversibly at constant pressure
- reversibly at constant pressure
- irreversibly at constant volume
- 18. For the same maximum pressure and temperature, what is the relation among the efficiencies of the Otto cycle, the Diesel cycle and the Dual cycle?

Mark only one oval.

ηDual > ηDiesel > ηOtto
 ηDiesel > ηDual > ηOtto
 ηDiesel > ηOtto > ηDual
 ηOtto > ηDiesel > ηDual

19. The coefficient performance of a refrigerator is 5. Calculate the temperature of the surrounding if the temperature inside the freezer is -20 degree C

- 11 degreeC
- 41 degreeC
- 21 degreeC
- 31 degreeC

20. The process will terminate when the pressure and temperature of the system and surrounding are equal. This state is referred as

Mark only one oval.

live state
no state.
open state.
dead state

21. function is defined as the difference between the internal energy and the product of temperature and entropy.

Mark only one oval.

\bigcirc	Helmhol	tz
\smile	i iciiiiioi	۲2

GIBB'S

📃 Kelvin

- Joule's Thompson
- 22.is expressed as the ratio of useful work Wu to the maximum useful work (Wu)max

Mark only one oval.

Irreversibility

Effectiveness

Availability

🔵 UE

23. A cyclic heat engine operates between a source temperature of 227 C and a sink temperature of 27 C. What will be the maximum efficiency of the heat engine?

Mark only one oval.

0100%

- 80%
- _____ 40%
- 60%
- 24. Carnot cycle is

Mark only one oval.

- reversible cycle
- _____ an irreversible cycle
- ____ practical cycle
- Dual cycle
- 25. COP of a heat pump is

- \bigcirc always less than infinity (COP < ∞)
- always less that 1 (COP < 1)
- _____ always equals to 1 (COP = 1)
- always equals to infinity (COP = ∞)

26. Which among the following is an example of a sink?

Mark only one oval.

Furnace where fuel is continuously burning

🔵 Sea

- Flywheel in running condition
- Compressed spring

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